

PHYS171L: General Physics II Lab

Fall 2014: August 25–December 17
F 2:00 PM–4:00 PM, Room: STB 209

Version 2: September 19, 2014 (subject to change)

Instructor: Kathy Cooksey, Ph.D.; STB 219; kcooksey@hawaii.edu; 808-932-7195

Office Hours: M 1–2 PM, T 12–1 PM, W 10–11 AM, and by appointment

Course Description: A laboratory supplement for PHYS107 and PHYS171; presents illustrative experiments in electricity, magnetism and optics.

Pre-requisites: PHYS107 or PHYS171 which can be taken concurrently.

Class Rules:

1. Students are required to read the lab manuals and view videos before the lab period and bring any pre-lab assignments to class (usually a table for data or a pre-formatted spreadsheet).
2. Students more than 15 minutes late to class will not be permitted to participate in that day's lab. Students with understandable issues with this (e.g., another class very far away, mobility problems, etc.) must discuss the situation with the instructor.
3. It is required for the students to conduct the lab experiment to submit a lab report. The lab reports are due in a week, whether there is a lab or not.
4. Students should be respectful and supportive of their peers' learning, which means helping each other with difficult concepts but not just giving the answer.

Email, Textbook, and Websites:

- UHH considers the `hawaii.edu` email and Lulima an official form of communication; students are responsible for receiving and returning information in a timely manner.
- There is no required textbook for the lab, but there are manuals and videos on the course Lulima site. At a minimum, there are two videos and one manual per lab; there are supplemental videos that show how various equipment are used, to be watched as needed.
- The Lulima course website is listed under PHYS-171L-004 (HIL.13319.FA14). This site will be the hub for all course information.

Lab Report Guidelines: The lab report should be typed and have a header and seven sections:

Header

Lab number, experiment title, student name, and lab partner's name (clearly distinguished) are at the top of the report.

1. Abstract

An abstract is a short summary of the entire lab, with emphasis on the results and conclusions.

2. Introduction

This section contains a brief overview of the objective of the lab. It is not be a word-for-word replication of the lab manual introduction.

3. Procedure

The procedure section is a brief re-stating—in the student's own words—of the steps and techniques required to complete the lab. It is important to also explain when and why the procedure deviated from the manual.

4. Raw Data

This section contains any data recorded during the course of the lab, which is usually presented in a table with appropriate units and uncertainties, if applicable. Partners' raw data must agree exactly.

5. Derived Data, Calculations, and Error Analysis

Here is where all calculations and any error analysis required by the lab manual go. If a calculation is repeated on many measurements, a complete example calculation is sufficient; rules for number of significant figures should be followed. The lab manual specifies how the final results should be presented (e.g., tables, charts, plots). Labels, units, and error bars should be included as appropriate.

6. Conclusion

The objective of the experiment is addressed here. This is done by analyzing the derived data, connecting it to the science objective, and justifying the significance based on the statistical analysis or percent errors, as required by the lab manual. Whether the results support the hypothesis must be directly addressed, including a comparison of the measured value(s) to the accepted one(s), as applicable. If the results do not support the hypothesis, a reasonable explanation is necessary. Data or calculations from the previous sections should be used to help make conclusions.

7. References

If any sources besides the lab manuals, videos, or course textbook are used, this is where they are cited. Citing a work does not mean it can be plagiarized.

Each student must submit an original lab report. Although raw data (numbers) are shared among lab partners, the calculations, error analysis, tables and figures, and conclusions are strictly an individual effort. See the example page at the end for a generic layout of a lab report.

Lab Schedule (subject to change):

Week	Dates	Lab
1	25–29 Aug	1. Electroscope
2	3–5 Sep	2. Electric Potential*
	1 Sep	<i>Labor Day (no class)</i>
3	8–12 Sep	3. Electric Fields
4	15–19 Sep	(No lab)
5	22–26 Sep	4. Kirchhoff's Laws [^]
6	29 Sep–3 Oct	5. RC Circuit*
7	6–10 Oct	(Make-up)
8	13–17 Oct	6. B near I *
9	20–24 Oct	7. Current Balance*
10	27–31 Oct	8. Faraday's Law
11	3–7 Nov	(Make-up)
	4 Nov	<i>Election Day (no class)</i>
12	10–14 Nov	9. RLC Circuit*
	11 Nov	<i>Veteran's Day (no class)</i>
13	17–21 Nov	10. Geometric Optics*
14	24–28 Nov	(No lab)
	27 Nov	<i>Thanksgiving Day (no class)</i>
	28 Nov	<i>Thanksgiving break (no class)</i>
15	1–5 Dec	11. Wave Optics
16	8–12 Dec	(Make-up)
17	15–19 Dec	<i>Finals Week</i>

* Indicate labs where the power supply supplemental video should be viewed.

[^] Indicate lab where DMM supplemental video should be viewed.

Grading:

- The grade depends on the average of the lab reports. Lab reports will be graded on a 10-point scale. Every part of the lab report is important but the emphasis is on the following:
 - logical conclusions and thoughtful discussions,
 - correct calculations and error analysis, and
 - clear tables and figures.
- One extra-credit point is possible, for example, for insightful knowledge of the lab topic or a particularly elegant and thoughtful way of presenting the data or figures.
- There are a limited number of opportunities to make up labs. Students may only make up labs for which they were excused.
 - If a student needs to miss a lab for a valid reason, s/he needs to contact the instructor before the start of class time; this includes official, University-related activities.
 - If a student were unable to email in advance due to extreme circumstances, s/he should contact the instructor as soon as possible.
- Late lab reports are accepted within 24 hours of the deadline for a maximum of 75% credit.
- The lowest grade of the completed lab reports will be dropped.
- As per the UHH Student Code of Conduct, cheating is not tolerated. Instances of cheating will be thoroughly investigated.
- The letter grade will be given based on the chart below:

Grade	% Required
A	≥ 93
A-	[90, 93)
B+	[87, 90)
B	[83, 87)
B-	[80, 83)
C+	[77, 80)
C	[73, 77)
C-	[70, 73)
D	[60, 70)
F	< 60

where e.g., [90, 93) means $\geq 90\%$ and $< 93\%$.

Disability Support: Any student with a documented disability who would like to request accommodation should contact the University Disability Services Office at 932-7623 (V) or 932-7002 (TTY), as early in the semester as possible.

Advising: Advising is a very important resource designed to help students complete the requirements of the University and their individual majors. Students should consult with their advisor at least once a semester to decide on courses, check progress towards graduation, and discuss career options and other educational opportunities provided by UH Hilo. Advising is a shared responsibility, but students have final responsibility for meeting degree requirements.

Kilohana Academic Success Center: The KASC provides academic support opportunities for all UH Hilo students that foster their development into independent, self-motivated learners. Students who visit Kilohana have access to subject-specific and academic skills tutoring from UHH students selected for their academic achievement and dedication to helping others succeed. Kilohana is located on the lower level of the Mookini Library and on the web at

<http://hilo.hawaii.edu/kilohana/>.

Human Rights: The University of Hawai'i at Hilo prohibits discrimination in its education programs based on race, national origin, color, creed, religion, sex, age, disability, veteran status, sexual orientation, gender identity or associational preference. If at any time during class you feel uncomfortable about what is being talked about, or feel that your human rights have been violated, please feel free to leave the room. However, the instructor asks that you confer with her as soon as possible about what happened so that appropriate action can be taken if necessary to avoid future problems. If you are uncomfortable speaking with the instructor about your concern, please contact Kalei Rapoza (kaleihii@hawaii.edu), Interim EEO/AA Director, at 932-7641.

UH Hilo Sexual Assault Policy: UH Hilo provides confidential assistance for victims of sexual assault. Counseling Services on-campus and the YWCA Sexual Support Services off-campus offer guidance regarding medical assistance and emotional help and can discuss options for reporting sexual assaults to law enforcement. All conversations are private and confidential. The UH Hilo Sexual Assault Policy can be found at: <http://hilo.hawaii.edu/uhh/vcsa/documents/UHHSexualAssaultPolicy.pdf> For assistance during the day, contact UH Hilo Counseling Services at (808) 932-7465; or, after hours and on weekends, contact the YWCA Sexual Assault Support Services at (808) 935-0677.

Student Conduct: Students are expected to follow the University of Hawai'i at Hilo Student Code of Conduct available at the following URL: <http://www.uhh.hawaii.edu/catalog/student-conduct-code.html>.

PHYS171L-004: Lab #1 – Electroscope

Student's Name
Lab Partner #1 Name
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1. Abstract

2. Introduction

3. Procedure

4. Raw Data

5. Derived Data, Calculations, and Error Analysis

6. Conclusion

7. References

(Your lab report may, of course, span more than one page.)